- A. F. Pozharskii, A. N. Suslov, and V. A. Kataev, Dokl. Akad. Nauk SSSR, <u>234</u>, 841 (1977).
- A. K. Sheinkman, T. S. Chmilenko, and T. M. Baranova, Khim. Geterotsikl. Soedin., No. 6, 849 (1982).

NEW METHOD FOR THE PREPARATION OF 2-(1-ADAMANTYL)THIOPHENE

V. P. Litvinov, V. I. Shvedov, and V. S. Dermugin

Among the numerous derivatives of adamantane [1] that have been of interest to researchers in the last 20 years owing to their high and diversified biological activity, compounds with a thiophene ring have received rather little study [2, 3]. This is due primarily to the lack of convenient methods for the synthesis of thienyladamantanes. The use of the alkylation of thiophene with bromo-substituted adamantanes in the presence of stannic chloride for this purpose gives a mixture of  $\alpha$ - and  $\beta$ -thienyladamantanes in a ratio of 2:1, from which one can isolate individual thienyladamantanes, although additional treatment of the mixture with mercuric chloride and subsequent hydrolysis of the resulting monochloromercuri derivatives are required for this [2].

We have developed a different method for the preparation of individual 2-(1-adamanty1)thiphene (III), which consists in construction of the thiophene ring on the basis of a functionally 1-substituted adamantane and is illustrated by the following scheme:



A 4.4-g sample of thioglycolic acid was added at 20°C to a solution of 7.5 g of  $\beta$ -(1-adamanty1)- $\beta$ -chlorovinyl aldehyde (II), obtained in 70% yield by Vilsmeier formylation of 1-acetyladamantane (I), in 30 ml of ethanol, after which 8.7 ml of triethylamine was added at 0-10°C, and the mixture was refluxed for 5 h. It was then worked up, the solvent was removed by distillation, and the residue was subjected to flash chromatography [with a 2 by 25 cm column filled with silica gel (40-100 mesh) by elution with CHCl<sub>3</sub>-heptane (1:1)] to give 2.67 g (34%) of 2-(1-adamanty1)thiophene (III) with mp 61.5-62°C (mp 62-62.5°C [2]). The aqueous layer, after extraction with ether, was acidified with concentrated HCl to give 0.4 g (4.5%) of 2-(1-adamanty1)thiophene-5-carboxylic acid (IV) with mp 267-269°C (dec., from benzene) (mp 266-268°C [3]). The compositions and structures of II-IV were confirmed by the results of elementary analysis and the IR, PMR, and mass spectra.

## LITERATURE CITED

R. C. Fort, The Chemistry of Diamond Molecules, Marcel Dekker, New York-Basel (1976).
W. Holk, J. Starting, and H. Wynberg, Rec. Trav. Chim., <u>85</u>, 1045 (1966).
W. Holk, H. Wynberg, and J. Starting, Rec. Trav. Chim., <u>85</u>, 1054 (1966).

N. D. Zelinskii Institute of Organic Chemistry, Academy of Sciences of the USSR, Moscow 117913. Translated from Khimiya Geterotsiklicheskikh Soedinenii, No. 8, pp. 1128-1129, August, 1982. Original article submitted February 8, 1982.

UDC 547.67:547.732